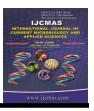


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## **Original Research Article**

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# **Evaluation of the Effectiveness of Live Synergistic Bacteria as a Bioremediation agent for selected Heavy Metals in Soils in Apapa-Lagos**

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#### ABSTRACT

#### Keywords

Bio-remediation, live synergistic bacteria, heavy metal, soil, contamination, sampling, analysis.

**Article Info** 

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Introduction

The high industrialization in Lagos state, Nigeria has made it one of the most heavily populated constituency on the earth. Lagos state was estimated to have a population of about 9.3millions in 2006, (NPC, 2006). The Lagos State Government estimates the population of Lagos at 17.5 million, although this number has been disputed by the Nigerian Government and found to be unreliable by the National Population Commission of Nigeria, which put the

The effectiveness of live synergistic bacteria as a bio-remediation agent for heavy metals such as arsenic, barium, cadmium, chromium, cobalt, copper, mercury, lead, nickel and zinc in soils in apapalagos was evaluated. Soil samples were taken from impacted stations and analysed for physicochemical parameters and heavy metals selected. Diluted live synergistic bacteria was applied on the site and a lag period was allowed. After the lag period, post application soil samples were again taken from the established sampling station and analysed for the selected parameters. Sampling and analysis were in accordance with standard procedures. The results show that live synergistic bacteria was able to adjust the soil pH, the electrical conductivity and total organic content but was not effective in adjusting heavy the metals for which it was tested within the limits of experimental errors.

> population at about 21 million in 2014. More than 50% of Nigeria's industrial activities including 300 industries in 12 industrial Estates are located in the Lagos area. The continuous increase in population and industrial growth in Lagos persistently cause large volume of waste to be generated (about 10,000 tons per day) (Oresanya, 2000) and environmentally safe landfills to cater for these wastes were inadequate. However, with the introduction of integrated

solid waste management system (ISWMS) initiated by the Lagos state waste management authority (LAWMA), there is significant reduction in the quantity of waste to be landfilled.

Apapa is one of the areas housing the industrial estates in Lagos including petroleum products tank farms and terminals. Soils in these areas are heavily contaminated as a result of the activities of these industries.

Usually, soils are considered as a sink for trace metals. Then, trace metals are able to move towards the water column or accumulate in plants and consequently contaminate the food chain (Odukoya et al., 2011). Of major concern about the presence of some metal ions in the environment are the negative health effects that they may cause in humans, animals, and plants (Mohammad and Nerges, 2009; Odukova and Abimbola, 2010). Metal distribution depends on the characteristics of the soils being studied and corresponds to the place of origin, such as the amount and type of organic and inorganic matter. Redox properties, pH and oxygen are among the most important chemical factors that affect the mobility of soil-bound metals [Duris, 2002; Guo and Zhou, 2006; Pandey et al., 2006 and Winfield, 2001].

of Anthropogenic metal sources contamination smelting include of electroplating, metalliferous ore, gas exhaust, energy and fuel production, application of fertilizers and municipal sludges to land and industrial manufacturing (Blaylock and Huang, 2000; Vasiliadou and Dordas, 2009, Keet al., 2001,

Studies of metal concentrations in soil ingested by people via the hand to mouth pathway have also been carried out in a

number of places according to Watt et al., 1993; Higgs et al., 1997. There is substantial evidence that a high Lead level in an environment could affect blood Lead level, intelligence and behavior (Bellinger et al., 1990; Lanphear et al., 1998). It is especially important that soil contents of potentially harmful substances are kept low in areas frequented by humans. Other metals such as Cd, Cu, Pd and Zn are good indicators of contamination in soils because they appear in gasoline, car component, oil lubricants and industrial incinerator emissions (Popoola et al., 2012). Trace amount of some metals such as trivalent chromium and cadmium entering the body via various routes can induce genetic and epigenetic alteration in different cancer related genes of somatic and stem cells, thus involving in cancer stem cell formation and increasing the incidence of cancer (Popoola et al., 2012).

Bioremediation is a waste management technique that involves the use of organisms to remove or neutralize pollutants from a contaminated site (NRC, 1993). Microorganisms are ideally suited to the task of contaminant destruction because they possess enzymes that allow them to use environmental contaminants as food and because they are so small that they are able to contact contaminants easily.

Whether microorganisms will be successful in destroying man-made contaminants in the subsurface depends on three factors: the type of organisms, the type of contaminant, and the geological and chemical conditions at the contaminated site (NRC, 1993). The goal bioremediation is to stimulate in microorganisms with nutrients and other chemicals that will enable them to destroy the contaminants. Although bioremediation currently is used commercially to cleanup a limited range of contaminants-mostly

hydrocarbons found in gasolinemicroorganisms have the capability to biodegrade almost all organic contaminants and many inorganic contaminants.

This work is to assess the effectiveness of a bioremediation agent – live synergistic bacteria, against selected heavy metals such as arsenic, barium, cobalt, cadmium, chromium, lead, copper, nickel and zinc in soils.

# Experimental

Soil samples were collected using hand trowel to the depth of contamination (0.5m). Composite samples were obtained from four spots at each visible impacted site and homogenized. The samples were taken for physico-chemical and heavy metals analysis in aluminium foil plates and stored in accordance with the provisions in the Environmental Guidelines And Standards Petroleum Industry for in Nigeria(EGASPIN, 2002) on the field before onward delivery to the laboratory (USEPA, 2000).

Soil with visible impacts (sheen, staining etc) were sampled and segmented into point 1, point 2, point 3 and point 4 as indicated in figure1.1.

Point 1 is approximately  $4.97m \times 4.0m = 19.88m2$ 

Point 2 is approximately  $1.74m \times 4.0m = 6.96m2$ 

Point 3 is approximately  $2.35m \times 4.0m = 9.40m2$ 

Point 4 is approximately  $10.28m \times 4.0m = 41.12m^2$ 

Laboratory analysis of the collected samples was guided by the statutory provisions in FMENV/DPR guidelines. The methodology for laboratory analyses was consistent with relevant established procedures (APHA, ASTM). Description of the methodology are presented below. Search gate laboratories limited, a government approved laboratory, was engaged for the analyses of the sample. The results of the analyses were recorded.

10% by volume of concentrated live synergistic bacteria was prepared using potable water. 230 litres of the 10% by volume of the live synergistic bacteria was applied to the selected impacted soil with the use of a watering can. The application took 10 hours to cover the area effectively. The area of the soil where the agent was applied was left and secured for 3 days to avoid further activities that could impact on the soil which could consequently weaken the effectiveness of the agent.

Another set of soil samples were taken from the four established sampling stations as "post remedial-agent application samples" using the same procedures. The second samples were analyzed for the previously indicated parameters and the results were recorded.

## **Results and Discussion**

The results of analyses conducted on the pre-application soil samples and those of the post application soil samples at the respective segmented points 1 to 4 are as shown in table 2.1.

From Table 2.1, the Ph values ranged from 5.1 to 5.8 during pre-application sampling while the pH values ranged from 5.7 to 6.1 during the post-application sampling. This indicates reduction in the acidity of the soil. Valuesbetween5and9suggestoptimalreductiv epathwayconditions.Songetal.(1990)reported similarresults with pH in a polluted soil after a bioremediation experiment. These behavior is represented in figure 2.1

Conductivity ranged from 26.9  $\mu$ S/cm to 214.4  $\mu$ S/cm during pre-application sampling while the value ranged from 27.3  $\mu$ S/cm to 210.8 $\mu$ S/cm after the lag phase. Values indicate anaerobic degradation trend. The behavior of conductivity in both the pre-application and post application sample is represented in a bar chart in figure 2.2.

Total organic content (TOC) ranged from 0.92 to 2.10 % during the pre-application sampling while the value ranged from 0.73 to 1.90 % after the lag phase. The result indicates that TOC has reduced. This suggest that biodegradation is likely occurring (Bragg *et al.*, 1994; Venosa *et al.*, 1996). As described in other studies, the degradation pattern of organic chemicals in soil usually shows a rapid initial phase of descent

followed by a period of little or no change in concentration. This kinetics is known as the "hockey stick" phenomenon (Alexander, 1999). The change of impacted soil total organic content is as shown in figure 2.3.

For heavy metals such as arsenic, barium, cadmium chromium, cobalt, copper, mercury, lead, nickel and zinc, the results are somewhat erratic. However, the general trend which is disenable shows that the agent was not effective in remediation against these metals. Barium, cadmium, chromium, cobalt, copper, lead, nickel and zinc were present in the samples. Lead recorded the highest concentration of 92.388mg/kg at pre-application point 4 and 83.052mg/kg at post application point 2.

### **Table.1** Soil samples handling and preservation procedure

Parameters	Containers	Preservative	Container pre- treatment
Metals: Ar, Ba, Cd, Cr, Co, Cu, Hg, Pb, Ni, Zn	Plastic bag	Freeze	Rinse with distilled water
Physico-chemical: TOC, pH, Conductivity	Plastic bag	Freeze	Rinse with distilled water

#### **Table.2** Laboratory analysis method

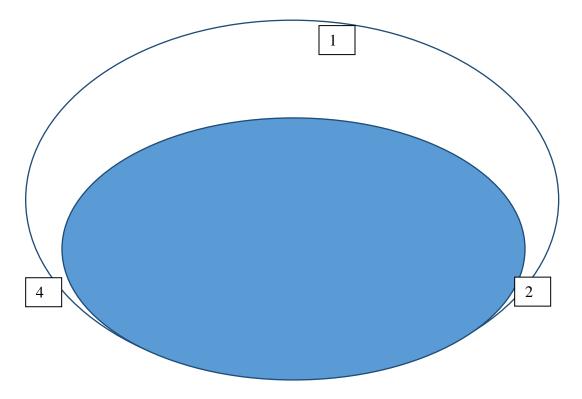
Parameter	Method	Method reference		
Total organic content (TOC)	Dichromate wet oxidation	USEPA 830		
Electrical conductivity	HACH Meter	Conductivity meter		
рН	Electrode	APHA 4500 H		
Heavy Metals: Cd, Cr, Co, Cu, Hg, Pb, Ni, Zn.	Atomic Absorption Spectrometry	APHA 3113 B		

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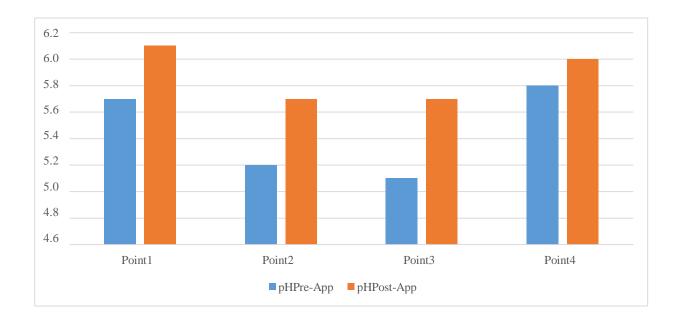
	PARAMETERS	Pre-Application			Post-Application			DPR Soil(mg/kg drymateri al)			
		Point1	Point2	Point3	Point4	Point1	Point2	Point3	Point4	Target Value	Intervention Value
	Physico-Chemical	vsico-Chemical					Physico-Chemical				
1	pН	5.7	5.2	5.1	5.8	6.1	5.7	5.7	6.		
2	Electrica 1	28.0	72.7	26.9	214.4	27.3	74.8	29.1	210.8		
	Conducti										
3	TOC(%)	0.98	1.30	0.92	2.10	0.73	0.99	0.87	1.90		
	HeavyMetals	avyMetals				HeavyMetals					
4	Arsenic(mg/kg)	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	29	55
5	Barium(mg/kg)	0.65	1.09	0.50	1.18	0.60	1.13	0.67	1.23	200	625
6	Cadmium(mg/kg)	0.573	0.382	0.763	0.954	0.490	0.321	0.793	0.271	0.8	12
7	Chromium(mg/kg)	0.05	0.05	< 0.001	0.05	< 0.001	0.05	0.06	0.07	100	380
8	Cobalt(mg/kg)	0.10	0.05	0.10	0.10	0.15	0.22	0.17	0.14	20	240
9	Copper(mg/kg)	1.609	2.574	2.252	2.574	2.736	1.568	2.526	5.451	36	190
10	Mercury(mg/kg)	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	< 0.001	0.3	10
11	Lead(mg/kg)	25.450	80.026	30.149	92.388	27.640	83.052	29.547	65.831	85	530
12	Nickel(mg/kg)	1.524	3.049	< 0.001	1.524	< 0.001	3.708	< 0.001	2.769	35	210
13	Zinc(mg/kg)	8.936	45.390	9.787	10.355	10.094	23.964	6.984	20.709	140	720

# **Table.3** Results of pre-application and post application soil samples.

# Fig.1 Layout of the impacted soil showing the four segments

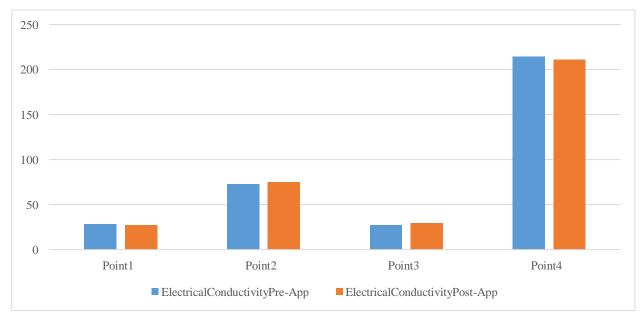


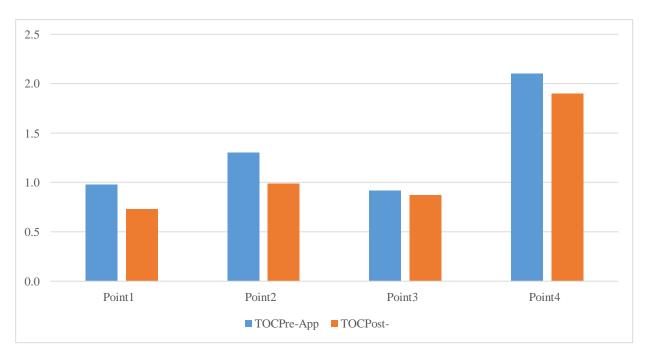
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# Fig.2 Change of impacted soil pH at the sample stations

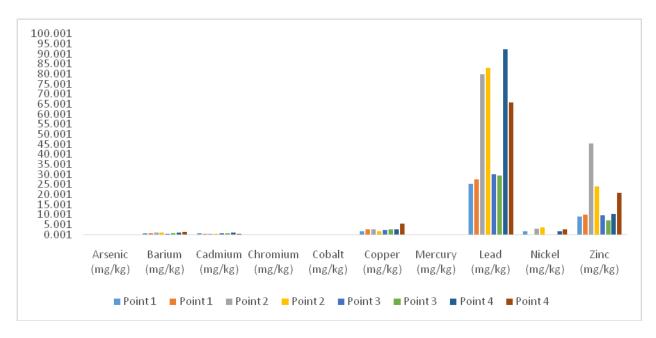








#### Fig.4 Change of Impacted Soil selected heavy metal content



Arsenic and have negligible mercury concentrations in the and pre post application soil samples. Though, the concentration of the metals in the soil samples were lower than the limits, one would expect a visible reduction in the

concentrations of these metals in the soil samples after application of the remedial agent, but this was not apparent in the results. The results are as shown in figure 2.4.

Another procedure could be to increase the number of lag days and then assess if there would be visible change in the concentration of these metal. Lack of materials have not made it possible for such procedure to be considered in this work.

The volume of impacted soil tested for remediation was approximately 38.68m<sup>3</sup>.

Approximately 1 gallon of 10% live synergistic bacteria was used for the remediation experiment of 1 cubic yard (0.765m<sup>3</sup>) of impacted soil. Based on the estimated volume of impacted soil, approximately 191.63 litres of the diluted solution was used on the site.

In conclusion, this work shows that impacted soil samples contain barium, cadmium, chromium, copper, lead, nickel and zinc at concentrations lower than the limits set by the department of petroleum resources. Lead recorded the highest concentration and chromium recorded the lowest concentration. It is also evident from the work that while the live synergistic bacteria could be effective in remediation at sites contaminated with organic compounds, its effectiveness against heavy metal is not visible within the period tested in this work. It is therefore recommended that further experiment should adjust the time lag between pre-application and post application analysis to really establish the potent of live synergistic bacteria in remediation actions at sites contaminated with heavy metals.

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